



Fabrication and Characterization of Lercanidipine Hydrochloride Solid Dispersions by Fusion Technique

Nazemoon Reddy^{1*}, Swarnalatha Dugasani² and Devanna Nayakanti³

^{1*} Research Scholar, Research and Development, Jawaharlal Nehru Technological University, Anantapur, Andhra Pradesh, India.

² Principal and Professor, Department of Pharmacognosy, Annamacharya College of Pharmacy, Rajampet, Kadapa, Andhra Pradesh, India.

³ Director, Jawaharlal Nehru Technological University-Oil Technological and Pharmaceutical Research Institute, Ananthapuramu, Andhra Pradesh, India

Abstract: The authors aimed to design solid dispersions with Lercanidipine Hydrochloride (LCD) with PVP K-30, Poloxamer-188, and HPMC K4M as carriers. Various mixtures of LCD and Polymers (PVP K-30, Poloxamer-188, and HPMC K4M) were made in 1:1, 1:3, 1:5 and 1:7 ratios, and the solid dispersion was prepared by melting tactic, later compressed into tablets. Drug excipient compatibility studies were examined by DSC and FTIR studies. LCD was found to be compatible with carriers used. The LCD solid dispersion was measured for physicochemical quality both in solid dispersions SD, and tablet states. The LCD solid dispersions found to have excellent flow possessions and compression assets. The yield of prepared solid dispersion was observed to be more than 90%), and the formulation LPOX-3 has showed a good yield of $98.9 \pm 1.95\%$. The tablets which were compressed from solid dispersions were found to have a uniform in size, shape, color, and consistency. The tablets were observed to have a uniform in thickness, and weight and ranged from 300.2 ± 1.64 to 301.7 ± 1.64 mg. The loss on friability was less than 1%, and the hardness was more than 4 Kg/cm² indicates significant mechanical strength and the LCD content was also found to be uniform (96.8 ± 1.35 to 99.9 ± 2.34). The solubility of LCD was found to be good in 0.1N HCl and diminished with an increase in pH of the buffer. LCD released from the tablets were firstly by eruption followed by zero order. The dissolution was found to be good in solid dispersions with LCD: Poloxamer-188 at the ratio of 1:5. The results obtained were satisfactory. The study concludes that LCD solid dispersions (LPOX-3) with 1:5 ratios of LCD and Poloxamer-188 was found to be a better carrier than PVP K-30, and HPMC K4M in increasing the solubility of LCD from the solid dispersions.

Keywords: Lercanidipine, Poloxamer-188, solid dispersions, solubility, dissolution.

*Corresponding Author

Nazemoon Reddy, Research Scholar, Research and Development, Jawaharlal Nehru Technological University, Andhra Pradesh, India.



Received On 28 April 2020

Revised On 23 May 2020

Accepted On 17 June 2020

Published On 05 January 2021

Funding This research did not receive any specific grant from any funding agencies in the public, commercial or not for profit sectors.

Citation Nazemoon Reddy^{1*}, Swarnalatha Dugasani², Devanna Nayakanti³, Fabrication and characterization of Lercanidipine Hydrochloride solid dispersions by fusion and solvent evaporation techniques..(2021).Int. J. Life Sci. Pharma Res.11(1), P78-86 <http://dx.doi.org/10.22376/ijpbs/lpr.2021.11.1.P78-86>

This article is under the CC BY- NC-ND Licence (<https://creativecommons.org/licenses/by-nc-nd/4.0>)
Copyright © International Journal of Life Science and Pharma Research, available at www.ijlpr.com



I. INTRODUCTION

The oral route is preferred as it is safe, effective, patient acceptance of all genders and age groups¹. The formulation aspect of solubility and dissolution are the main issues for drugs with poor solubility². The researchers do various trials in elevating the solubility of such drugs. Several methodologies adopted to uplift the drug solubility viz., co-solvency, complexation, salt formation, adding surfactants, micronization, and pH alteration etc., amongst solid dispersion (SD) methodology situated on the top priority³, for its ease, modest, and resourceful scheme in amassing the solubility. Lercanidipine hydrochloride (LCD) is prescribed for its calcium channel blocking activity, it is a BCS class II drug with t_{1/2} of 8h and bioavailability of 10%⁴. LCD is prescribed for hypertension and angina patients. The poor aqueous solubility of LCD, which restricts the onset of action^{5,6}. Various polymers were tried as carriers for solid dispersions like Sorbitol, Mannitol, Citric acid, Succinic acid, Polyvinyl Pyrrolidones, Polyethylene Glycols, cellulose derivative, and Eudragits etc., Literature review revealed that

many attempts have been tried for making solid dispersions using the carriers used in the study, but no attempts have been made in combination of these carriers (PVP K-30, Poloxamer-188, and HHPMC K4M.). So, the scholars made an effort in appraising the LCD solubility by SD made by melting using Poly Vinyl Pyrrolidone (PVP) K-30, Poloxamer-188, and Hydroxy Propyl Methyl Cellulose (HPMC) K4M.

2. MATERIALS AND METHODS

2.1. Materials

The LCD was gifted from Torrent Pharmaceuticals, Mumbai. PVP K-30, Poloxamer-188, HPMC K4M, Microcrystalline Cellulose, Talc, and Magnesium stearate were procured from SD Fine chemicals India. Double distilled water was used when needed.

2.2. Scheming of Solid dispersions

The numerous plans of LCD, SD were exemplified in table 1.

Drug: Carrier	Drug: Carrier ratio	Formulation code
LCD: PVP K30	1:1	LPVP-1
	1:3	LPVP-2
	1:5	LPVP-3
	1:7	LPVP-4
LCD: Poloxamer 188	1:1	LPOX-1
	1:3	LPOX-2
	1:5	LPOX-3
	1:7	LPOX-4
LCD: HPMC K4M	1:1	LHPM-1
	1:3	LHPM-2
	1:5	LHPM-3
	1:7	LHPM-4

The polymers were melted based on geometric melting from higher to lower i.e., HPMC-K4M, PVP K-30, then Poloxamer-188 in a ceramic dish. LCD was spread in the molded mass with constant amalgamation. The blend was congealed at room temperature^{7,8}. The SD were stored in a desiccator (ABG Initiatives, Hyderabad, Telangana) for a day, slightly furrowed in a mortar (Aruna Scientific, Hyderabad, Telangana). The formed SD were passed through # 30 mesh

(ASTM E 11, Hyderabad, Telangana) to get uniform sized subdivisions.

2.3. Fabricating of solid dispersion tablets

The SD corresponding to 20 mg of LCD were prepared after combination⁹ with components (as per table 2) compressed in the 8 station tablet compression machine (Karnavati, India).

Ingredients	Quantity per tablet (mg)
Solid dispersions equivalent to 20 mg of LCD	150
Lactose	75
Starch	15
Micro Crystalline Cellulose	50
Magnesium stearate	5
Talc	5
Weight of the tablets	300

2.4. Evaluations

2.4.1. Melting point

The crystalline chemicals and drugs are available as pure form and have sharp melting points¹⁰. The preliminary evaluation is the determination of the LCD melting point using the melting point apparatus (MT-934, Mumbai). The melting temperature of the LCD was recorded three times.

2.4.2. Solubility studies

LCD pure drug was examined for solubility in 0.1N HCl, water, pH 4.5 Acetate buffer, pH 6.8 and pH 7.4 Phosphate buffers¹¹.

2.4.3. Drug-excipients compatibility studies

The DSC and FTIR studies were made to check the compatibility amongst the LCD and the carriers used in making SD.

2.4.3.1. Differential Scanning Calorimetry (DSC)

A 1:1 ratio of LCD, and ~20mg of Polymer (PVP K30/Poloxamer 188/HPMC K4M), were placed in DSC

crucible and heated from till 500°C in DSC apparatus (DSC-50, Shimadzu, Japan).

2.4.3.2. Fourier-transform infrared (FTIR) spectroscopic study

The dealings among constituents of the SD were established using scanning in FTIR spectroscopy. The FTIR spectra of the LCD, with combination with, were renowned by the FTIR spectrometer (Bruker) by scanning at 4000-400 cm⁻¹.

2.5. Evaluations of LCD Solid Dispersions

The gained SD were scrutinized for the cited strictures.

2.5.1. Flow properties

The SD were assessed for flow restraints viz., angle of repose, densities, Carr's Index, and Hausner's ratio^{12,13}.

2.5.2. Yield

The weight of dried SD to the total weight of ingredients used in making SD can be assessed by the formula given¹⁴.

$$\% \text{ Yield} = \frac{\text{Actual weight of the SD}}{\text{Total weight of drug and excipients}} \times 100$$

2.6. Depiction of tablets made with SD

The SD were compressed into tablets and were measured for the following properties.

2.6.1. Uniformity in size and shape.

The SD tablets were inspected under a dissection microscope (DM-100, Mumbai) for their size and shape¹⁵.

2.6.2. Thickness

The tablets were held between Vernier Caliper's (Qumos Enterprises, India) jaws and breadth was assessed 3 times¹⁶.

2.6.3. Uniformity in weight

20 tablets from each batch of tablets were distinctly weighed

with an electronic digital balance (Citizen, CY-104, Mumbai, India) and the average was assessed. The nonconformity was matched with IP limits ($\pm 5\%$ for 300 mg tablet)¹⁷.

2.6.4. Hardness

The tablets were pushed between the two extremes of Pfizer tablet hardness tester. The energy to break the tablets was performed 3 times¹⁸.

2.6.5. Friability

Surface abrasion may emerge while tablet handling can be assessed by a Roche Friabilator. Initially weighed tablets (10 tablets), were allowed to fall from a height of 6 inches for 100 resolutions, the tablets were de-dusted and weighed again. The loss in weight was assessed by the formula given¹⁹.

$$F = \frac{W_{\text{initial}} - W_{\text{final}}}{W_{\text{initial}}} \times 100$$

2.6.6. Calibration curve

100 mg of LCD dissolved in pH 1.2, HCl solution (0.1M). Dilutions viz., 2, 4, 6, 8 and 10 $\mu\text{g}/\text{mL}$ were prepared and scanned spectrophotometrically at 239 nm then the absorbance verses concentrations produces the calibration curve of LCD²⁰.

2.6.7. Uniformity of drug content

10 tablets were pulverized. A blend equivalent to 20 mg LCD

was dissolved in methanol, diluted and the absorbance was measured at λ_{max} of 239 nm²¹.

2.6.8. In-vitro drug release studies

The USP paddle apparatus containing 0.1N HCl (900mL), stirred at 50 rpm and retained at $37 \pm 0.5^\circ\text{C}$. The media was withdrawn at regular intervals for 1h, filtered using Whatman filter paper and diluted to 10 mL with 0.1N HCl, and analyzed at λ_{max} of 239 nm by UV/visible spectrophotometer²². The release data was further kinetically

assessed by zero-order, first-order²³, and Hixson Crowell's models²⁴.

2.7. Scanning Electron Microscopy

The surface topography of SD was confirmed by scanning the surface of SD by scanning electron microscopy²⁵ (Perkin Elmer, USA). An accelerating voltage of 20KV was used and the images obtained at the magnification of ×500.

3. RESULTS AND DISCUSSION

LCD melts at 197.5±1.29°C, designates the purity of the LCD (as it melts in between 196-198°C). The LCD presented good solubility in 0.1N HCl (0.313±0.01 µg/mL) relatively in Water, Acetate buffer (pH4.5), Phosphate buffer (pH6.8) and Phosphate buffer (pH7.4). The solubility data for pure LCD was illustrated in figure 1.

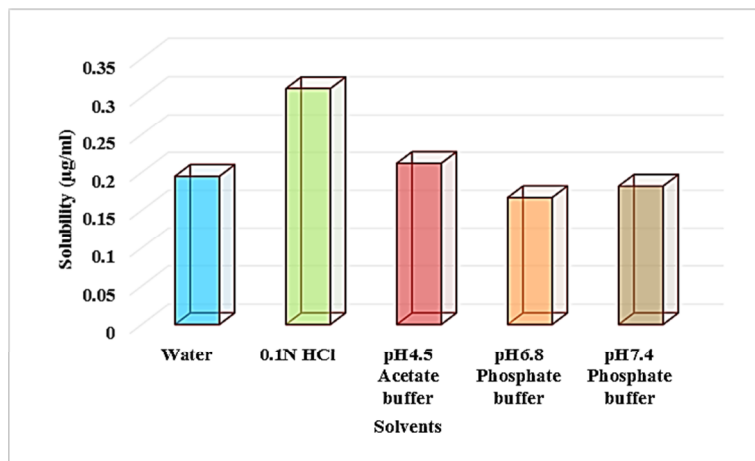


Fig 1. The solubility of pure LCD in various solvents

The DSC thermograms of LCD with PVP K30, Poloxamer 188/HPMC K4M carriers were moved to the lesser

temperatures representing certain associating of LCD with carriers adopted (figure 2).

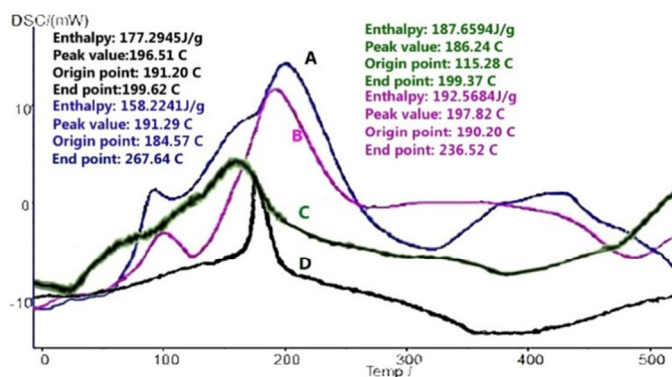


Fig 2. DSC thermograms of LCD (A) Pure drug (B) with PVP K30 (C) with Poloxamer 188 (D) with HPMC K4M

The FTIR study revealed that the distinctive peaks and stretches of LCD pure drugs were also found in LCD – carriers designate no negative discordancy of LCD with

carriers used. The FTIR spectra of LCD pure and carriers were shown in figure 3.

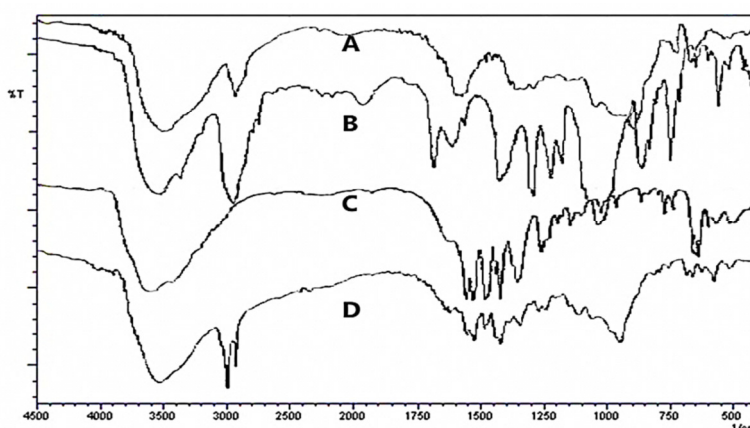


Fig 3. FTIR spectra of LCD (A) Pure drug (B) with PVP K30 (C) with Poloxamer 188 (D) with HPMC K4M

When the LCD-SD assessed for the angle of repose was found to be 25 to 30° i.e., 24.95±0.01 to 29.65±0.02°, which authorizes excellent flow possessions. On the other hand, the compressibility Index was less than 10 (2.935 to 7.978)

and Hausner ratio less than 1.09 (1.030 to 1.086), demonstrating good compression assets while tableting. The flow properties of LCD-SD were briefed in table 3.

Formulation	Flow properties				
	Angle of repose (°)	Bulk Density	Tapped Density	Carr's Index	Hausner Ratio
LPVP-1	27.84±0.01	0.529	0.545	2.935	1.030
LPVP-2	26.25±0.02	0.569	0.598	4.849	1.050
LPVP-3	25.84±0.03	0.658	0.678	2.949	1.030
LPVP-4	24.95±0.01	0.628	0.648	3.086	1.031
LPOX-1	26.65±0.02	0.458	0.487	5.954	1.063
LPOX-2	25.29±0.06	0.518	0.537	3.538	1.036
LPOX-3	26.21±0.08	0.635	0.665	4.511	1.047
LPOX-4	29.23±0.05	0.648	0.698	7.163	1.077
LHPM-1	27.10±0.07	0.569	0.598	4.849	1.050
LHPM-2	28.45±0.05	0.519	0.564	7.978	1.086
LHPM-3	29.65±0.02	0.587	0.622	5.627	1.059
LHPM-4	26.98±0.04	0.498	0.534	6.741	1.072

Values in mean ±SD; trials made (n=3)

The yield of LCD-SD was observed to be good (>90%), and LPOX-3 has a good yield of 98.9±1.95%. The LCD-SD tablets were seeming to have a uniform in size, shape, pale white-colored, odorless with a smooth surface. The tablets were found to have a uniform in thickness, ranged from 4.50±0.01 to 4.52±0.04 mm, and weight and ranged from 300.2±1.64 to

301.7±1.64 mg. The loss on friability was between 0.15±0.02 to 0.84±0.02%, which is < 1%, and the hardness was ranged from 5.8±0.05 to 8.7±0.02 (>4 Kg/cm²) representing that the tablets bearing significant mechanical strength and the LCD content was also found to be uniform (96.8±1.35 to 99.9±2.34). All these values were explained in table 4.

Formulation	Physical parameter					
	Uniformity of Weight (mg)	Hardness (cm ²)	Thickness (mm)	Friability (%)	Yield (%)	Assay (%)
LPVP-1	300.2±3.29	7.9±0.06	4.51±0.02	0.84±0.02	97.1±1.20	96.8±1.35
LPVP-2	301.0±1.27	8.4±0.05	4.50±0.03	0.68±0.02	98.0±3.28	97.9±1.95
LPVP-3	301.2±2.38	6.1±0.03	4.51±0.06	0.48±0.01	98.3±1.54	98.8±2.35
LPVP-4	301.3±1.39	8.7±0.02	4.52±0.04	0.32±0.02	97.6±2.35	97.9±4.25
LPOX-1	301.1±3.25	6.5±0.01	4.51±0.02	0.49±0.02	96.8±3.16	96.9±1.25
LPOX-2	300.2±1.64	7.2±0.03	4.50±0.01	0.15±0.02	95.2±1.37	98.4±3.02
LPOX-3	301.2±2.35	7.1±0.01	4.50±0.01	0.54±0.02	98.9±1.95	97.7±2.20
LPOX-4	301.5±1.68	6.8±0.02	4.50±0.02	0.35±0.03	98.5±3.16	98.4±1.62
LHPM-1	301.6±2.36	6.3±0.01	4.51±0.01	0.62±0.02	96.5±1.25	98.5±2.31
LHPM-2	300.3±1.28	7.2±0.05	4.50±0.06	0.18±0.01	96.3±1.39	97.2±1.24
LHPM-3	301.7±1.64	5.8±0.05	4.52±0.03	0.44±0.02	98.2±3.26	99.7±3.25
LHPM-4	301.6±1.39	7.4±0.02	4.51±0.04	0.39±0.01	96.4±2.48	99.9±2.34

Values in mean ±SD; trials made (n=3)

The solubility of LCD was found to be good in 0.1N HCl and diminished with an increase in pH of the buffer. Among them,

LPOX-4 signified good solubility in 0.1 N HCl. The entire description of solubility was embodied in figure 4.

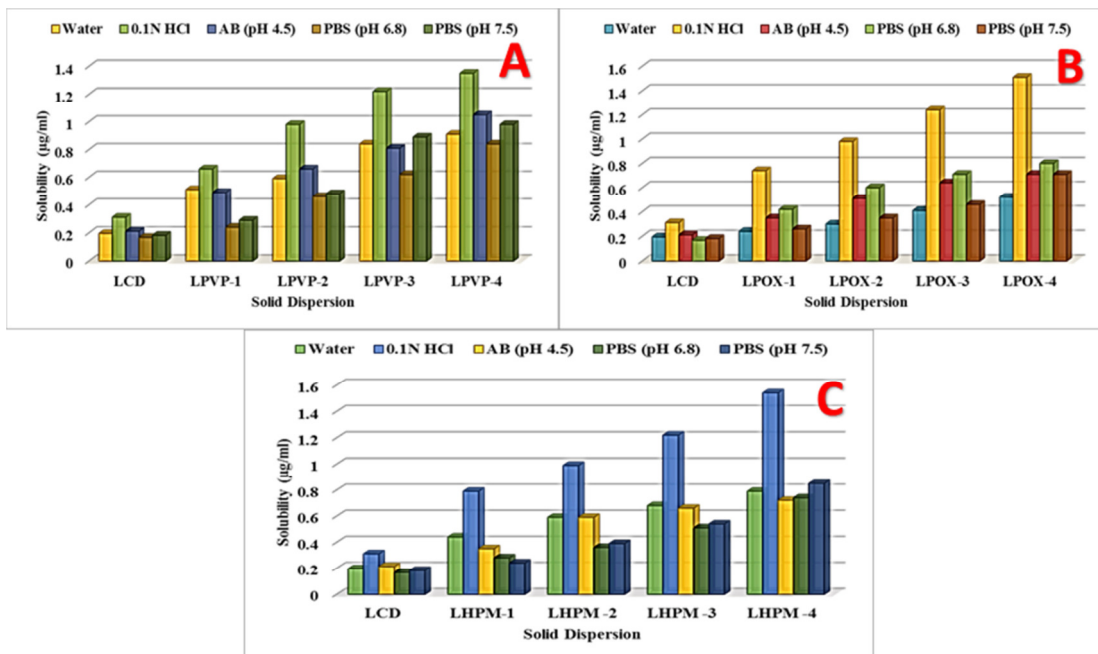


Fig 4. Solubility details of Lercanidipine solid dispersions prepared with A) PVP K30 B) Poloxamer-I88 C) HPMC K4M

LCD followed Beer’s Lambert’s law at 2 to 10 µg/mL. The regression (R^2 value was detected to be 0.9992 with a slope of $0.0749x+0.0157$). The LCD was determined by plotting the calibration curve of the LCD. LCD released from the tablets were firstly by eruption < 10 min and the end of 1h the LCD

was released in zero order. The dissolution of prepared tablets was found good in SD with LCD: Poloxamer-I88 at the ratio of 1:5 (figure 5), which followed zero order. The kinetic study revealed that the LCD-SD followed first-order release kinetics and illustrated in figures 6 and 7.

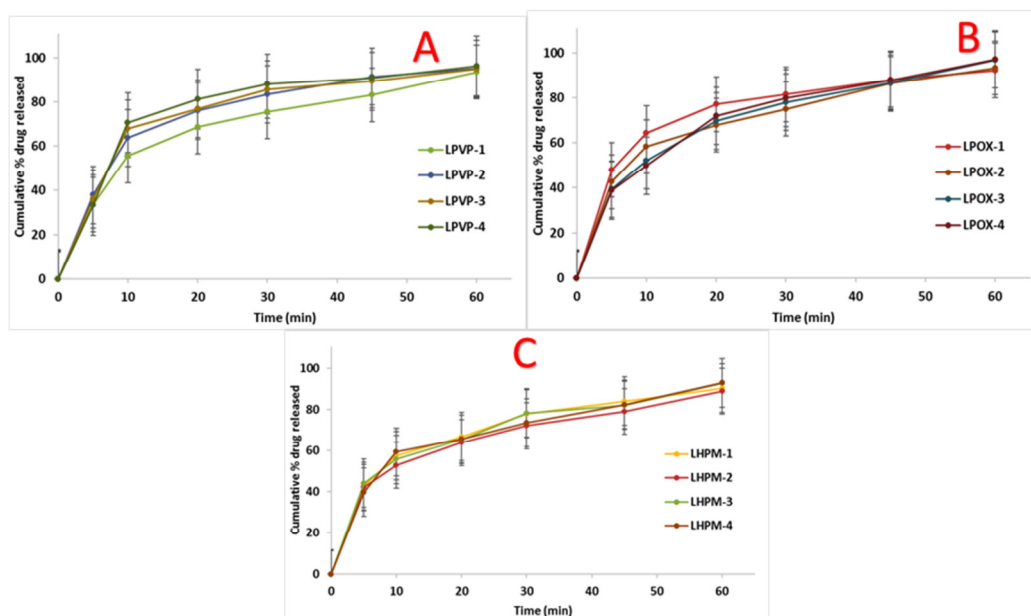


Fig 5. In vitro dissolution profile of LCD with A) PVP K30 B) Poloxamer-I88 C) HPMC K4M

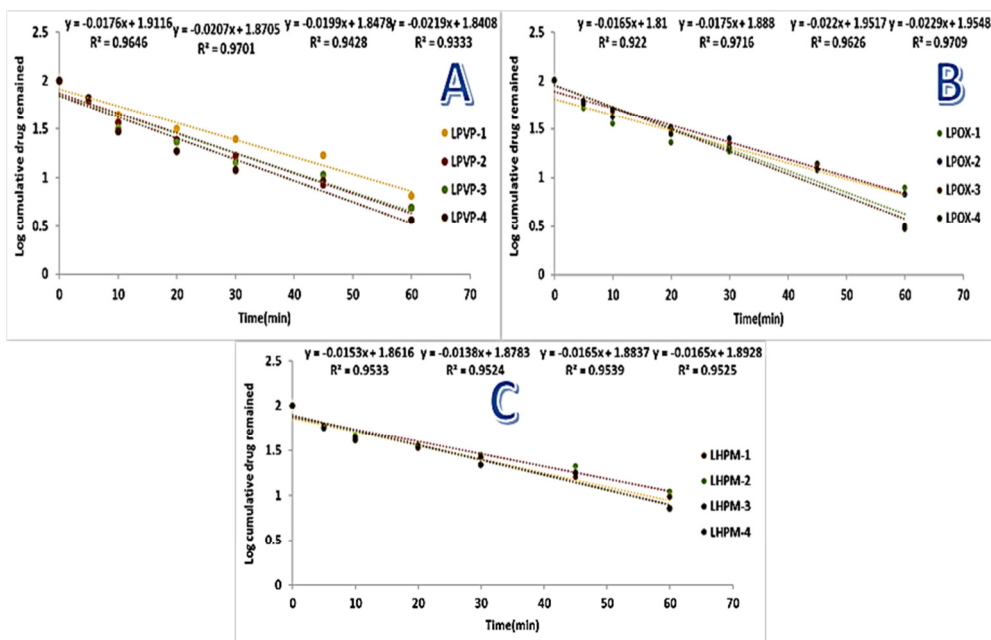


Fig 6. First-order release kinetics of LCD with A) PVP K30 B) Poloxamer-I88 C) HPMC K4M

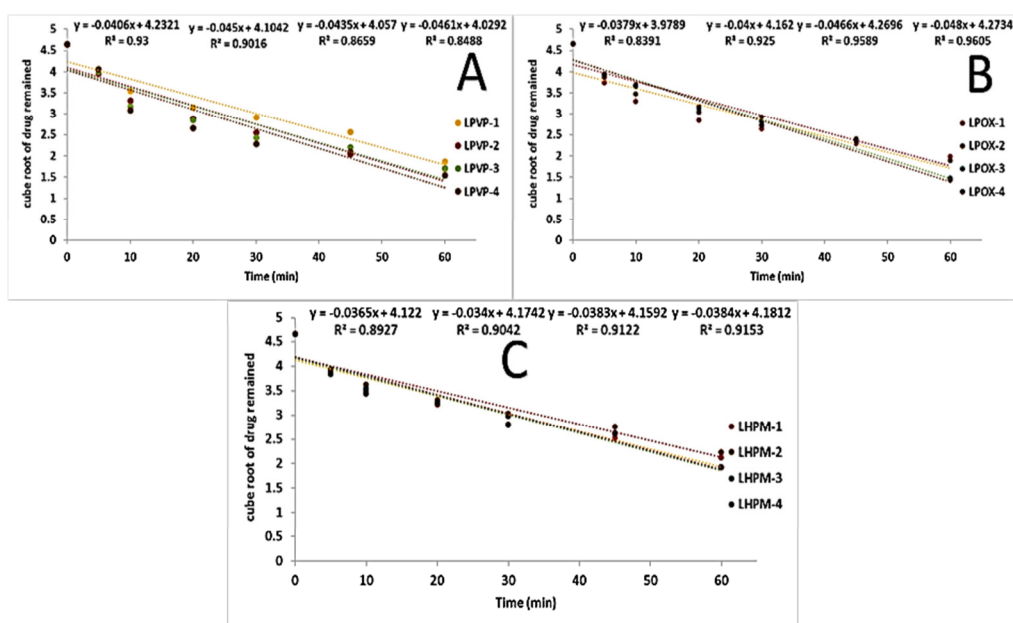


Fig 7. Hixson Crowell's plots of LCD with A) PVP K30 B) Poloxamer-I88 C) HPMC K4M

The SEM analysis revealed that SD with PVP K30 and Poloxamer-188 produce an amorphous SD. In the case of Poloxamer-188, which acts as a crystal inhibitor, this may be

the reason for the enhancement of dissolution. The SEM analysis images were represented in figure 8.

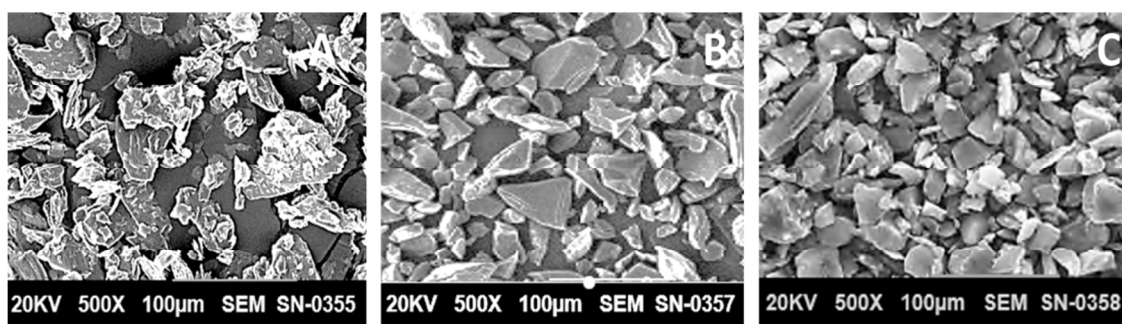


Fig 8. SEM analysis of LCD-SD with A) PVP K30 B) Poloxamer-I88 C) HPMC K4M

LCD was found to show elevated solubility in 0.1M HCl compared to its pure LCD. The angle of repose of LCD-SD

represented excellent flow properties, additionally, the compressibility Index and Hausner ratio proved the good

compression assets of prepared SD. The yield of LCD-SD was found enhanced (up to $98.9 \pm 1.95\%$) compared to other approaches using PVP K-30.²⁶ The LCD-SD tablets were found to have uniformity in physicochemical constraints including the loss on friability was below 1% with $>4 \text{ Kg/cm}^2$ hardness, and uniformity in LCD drug content. This rapid dissolution needed to assist in enhancing the release of LCD from the SD.²⁷ The prepared SD showed good LCD release within 10 min, which might be due to the solubility enhancing stuff of Poloxamer-188 when combined with LCD.²⁸ The release rate was significantly increased when the LCD: Poloxamer-188 ratio was at 1:5. Similar observations were also reported by Shamsuddin et al.²⁹ The LCD release from all the SD followed first-order kinetics, as the plot observed in between log percentage drug remaining versus time was found to be linear with a coefficient of correlation ($R^2 = 0.9964$). The correlation coefficient (r) values of the first-order release model are found to be 0.9912-0.9964, which is slightly higher compared to the Hixson-Crowell's cube root model. Hence, the release of drug from the SD followed mainly first-order kinetics compared to the Hixson-Crowell cube root law. The SEM analysis revealed that SD with PVP

K30 and Poloxamer-188 produce an amorphous SD. this may be the reason for the enhancement of dissolution.

4. CONCLUSION

The study discovered that the solid dispersions prepared by Poloxamer-188 were good carriers for elevating the solubility of Lercanidipine by making solid dispersions. The LPOX-4 formulation with 1: 5 proportions of Lercanidipine and Poloxamer-188 made by the melting methodology were good in elevation of *in vitro* dissolution of Lercanidipine and it followed first-order release kinetics.

5. AUTHORS CONTRIBUTION STATEMENT

Nazmoon Reddy conceived the presented idea, developed the theory and performed the computations. Swarnalatha Dugasani, and Devanna Nayakanti verified and corrected the manuscript. All authors discussed the results and contributed to the final manuscript.

6. CONFLICTS OF INTEREST

Conflicts of interest declared none.

7. REFERENCES

- Varadarajulu S, Tamhane A, Drelichman ER. Patient perception of natural orifice transluminal endoscopic surgery as a technique for cholecystectomy. *Gastrointestinal endoscopy*. 2008 May 1;67(6):854-60 doi: 10.1016/j.gie.2007.09.053
- Van den Mooter G. The use of amorphous solid dispersions: A formulation strategy to overcome poor solubility and dissolution rate. *Drug Discovery Today: Technologies*. 2012 Jun 1;9(2): e79-85. doi: 10.1016/j.ddtec.2011.10.002
- Annepogu H, Ahad HA, Nayakanti D. Assessing the best poly vinyl pyrrolidone as a carrier for Etoricoxib solid dispersions: fabrication and evaluation. *J. Pharm. Sci. Innov.* 2018; 7(5), 208-2014. doi: 10.7897/2277-4572.075109
- Ramasahayam B, Eedara BB, Kandadi P, Jukanti R, Bandari S. Development of isradipine loaded self-nano emulsifying powders for improved oral delivery: in vitro and in vivo evaluation. *Drug development and industrial pharmacy*. 2015 May 4;41(5):753-63. doi: 10.3109/03639045.2014.900081
- Dokania S, Joshi AK. Self-microemulsifying drug delivery system (SMEDDS)—challenges and road ahead. *Drug delivery*. 2015 Aug 18;22(6):675-90. DOI: 10.3109/10717544.2014.896058
- Hearnden V, Sankar V, Hull K, Juras DV, Greenberg M, Kerr AR, Lockhart PB, Patton LL, Porter S, Thornhill MH. New developments and opportunities in oral mucosal drug delivery for local and systemic disease. *Advanced drug delivery reviews*. 2012 Jan 1;64(1):16-28. doi: 10.1016/j.addr.2011.02.008
- Dong Z, Chatterji A, Sandhu H, Choi DS, Chokshi H, Shah N. Evaluation of solid state properties of solid dispersions prepared by hot-melt extrusion and solvent co-precipitation. *International journal of pharmaceutics*. 2008 May 1;355(1-2):141-9. doi: 10.1016/j.ijpharm.2007.12.017
- Agrawal AM, Dudhedia MS, Patel AD, Raikes MS. Characterization and performance assessment of solid dispersions prepared by hot melt extrusion and spray drying process. *International journal of pharmaceutics*. 2013 Nov 30;457(1):71-81. doi: 10.1016/j.ijpharm.2013.08.081
- Gupta MK, Goldman D, Bogner RH, Tseng YC. Enhanced drug dissolution and bulk properties of solid dispersions granulated with a surface adsorbent. *Pharmaceutical development and technology*. 2001 Jan 1;6(4):563-72. doi: 10.1081/PDT-120000294
- Marsac PJ, Li T, Taylor LS. Estimation of drug-polymer miscibility and solubility in amorphous solid dispersions using experimentally determined interaction parameters. *Pharmaceutical research*. 2009 Jan 1;26(1):139. doi: 10.1007/s11095-008-9721-1
- Badawy SI, Hussain MA. Microenvironmental pH modulation in solid dosage forms. *Journal of pharmaceutical sciences*. 2007 May 1;96(5):948-59. doi: 10.1002/jps.20932
- Miller RL, Byrne RJ. The angle of repose for a single grain on a fixed rough bed. *Sedimentology*. 1966 Jul;6(4):303-14. doi: 10.1111/j.1365-3091.1966.tb01897.x
- Ileleji KE, Zhou B. The angle of repose of bulk corn stover particles. *Powder technology*. 2008 Oct 28;187(2):110-8. doi: 10.1016/j.powtec.2008.01.029
- Ghareeb MM, Abdulrasool AA, Hussein AA, Noordin MI. Kneading technique for preparation of binary solid dispersion of meloxicam with poloxamer 188. *Aaps Pharmscitech*. 2009 Dec 1;10(4):1206-15. doi: 10.1208/s12249-009-9316-0
- McMaster PD, Parsons RJ. The effect of the pulse on the spread of substances through tissues. *The Journal of experimental medicine*. 1938 Aug 31;68(3):377. doi: 10.1084/jem.68.3.377
- Lohman TG, Pollock ML. Skinfold measurement: Which caliper? How much training? *Journal of physical education and recreation*. 1981 Jan 1;52(1):27-9.

- doi: 10.1080/00971170.1981.10629017
17. Pharmacopoeia I. Uniformity of weight of single-dose preparations. Ghaziabad: The Indian Pharmacopoeia Commission, Central Indian Pharmacopoeia Laboratory, Govt. of India, Ministry of Health & Family Welfare. 2008:182.
 18. Fairchild HJ, Michel F. Pfizer tablet hardness tester. *Journal of pharmaceutical sciences*. 1961 Nov;50(11):966-9. doi: 10.1002/jps.2600501119
 19. Shafer EG, Wollish EG, Engel CE. The "Roche" friabilator. *Journal of the American Pharmaceutical Association*. 1956 Feb;45(2):114-6. doi: 10.1002/jps.3030450214
 20. Shaikh F, Patel V, Patel M, Surti N. Dissolution Method Development and Validation for Lercanidipine Hydrochloride Tablets. *Dissolution Technologies*. 2018 Feb 1;25(1):38-46. doi: 10.14227/DT250118P38
 21. Jang DJ, Bae SK, Oh E. Coated dextrin microcapsules of amlodipine incorporated into orally disintegrating tablets for geriatric patients. *Biomedicine & Pharmacotherapy*. 2014 Oct 1;68(8):1117-24. doi: 10.1016/j.biopha.2014.10.010
 22. Kallakunta V, Bandari S, Jukanti R, Veerareddy P. Formulation and evaluation of oral self-emulsifying powder of lercanidipine hydrochloride. *Powder Technol*. 2012; 221:375-82. doi: 10.1016/j.powtec.2012.01.032
 23. Havlin JL, Westfall DG, Olsen SR. Mathematical models for potassium release kinetics in calcareous soils I. *Soil Science Society of America Journal*. 1985;49(2):371-6. doi: 10.2136/sssaj1985.03615995004900020020x
 24. Karasulu E, Karasulu HY, Ertan G, Kirilmaz L, Güneri T. Extended release lipophilic indomethacin microspheres: formulation factors and mathematical equations fitted drug release rates. *European journal of pharmaceutical sciences*. 2003 Jun 1;19(2-3):99-104. doi: 10.1016/S0928-0987(03)00048-4
 25. Falconer JR, Wen J, Zargar-Shoshtari S, Chen JJ, Mohammed F, Chan J, Alany RG. The effects of supercritical carbon dioxide processing on progesterone dispersion systems: a multivariate study. *AAPS PharmSciTech*. 2012 Dec 1;13(4):1255-65. doi:10.1208/s12249-012-9850-z
 26. Chinchawade Ashlesha B, Gadhave Manoj V. Enhancement of dissolution rate of lercanidipine by solid dispersion technique. *World Journal of Pharmaceutical Research*, 2015; 4(2): 1192-1199.
 27. Shah NH, Carvajal MT, Patel CI, Infeld MH, Malick AW. Self-emulsifying drug delivery systems (SEDDS) with polyglycolized glycerides for improving in vitro dissolution and oral absorption of lipophilic drugs. *International journal of pharmaceutics*. 1994 May 16;106(1):15-23. doi: 10.1016/0378-5173(94)90271-2
 28. Hallan SS, Kaur P, Kaur V, Mishra N, Vaidya B. Lipid polymer hybrid as emerging tool in nanocarriers for oral drug delivery. *Artificial cells, nanomedicine, and biotechnology*. 2016 Jan 2;44(1):334-49. doi: 10.3109/21691401.2014.951721
 29. Shamsuddin MF, Ansari SH, Ali J. Development and evaluation of solid dispersion of spironolactone using fusion method. *International journal of pharmaceutical investigation*. 2016 Jan;6(1):63. doi: 10.4103/2230-973X.176490.